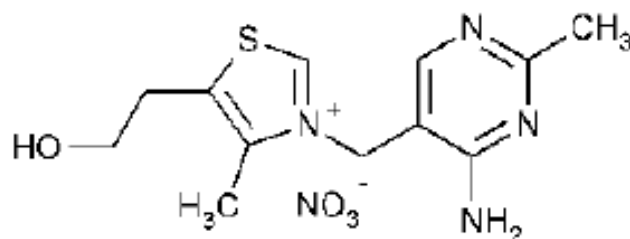




Houston Community College

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SAMPLE EXAM # 1B
Organic Chemistry I
2425



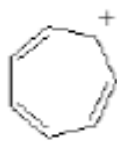
Thiamin

Thiamin— Thiamine Mononitrate: Nitrate salt form of thiamine for the preparation of various multivitamins. Thiamine Pyrophosphate: The coenzyme in a variety of reactions or prosthetic component of carboxylase.

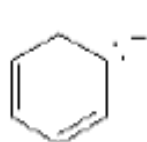
CHEM 2425 - Sample Exam #2A (Chapters 13, 14, and 15)

Part I. Multiple choice questions. Please write your answer in the space provided.

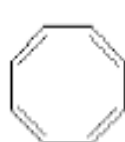
____ 1. Which of the following is aromatic?



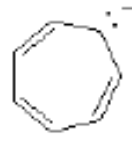
A.



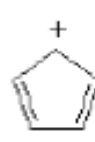
B.



C.



D.



E.

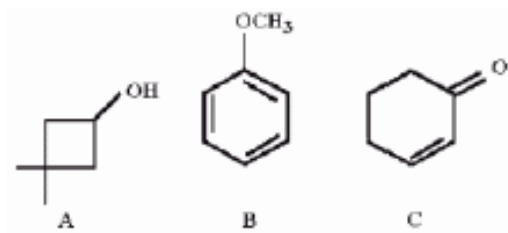
____ 2. What is the major difference between an antiaromatic and aromatic compound?

- A) The structure must be cyclic for aromatic but not antiaromatic compounds?
- B) Antiaromatic compounds have at least one sp^3 hybridized atom in the ring.
- C) Antiaromatic compounds can assume a chair-like structure while aromatic compounds are nearly flat.
- D) Aromatic compounds cannot have a charged atom in the structure.
- E) Only aromatic compounds follow Huckle's rule.

____ 3. 2-Methylhexane shows an intense peak in the mass spectrum at $m/z = 43$. Propose a likely structure for this fragment.

- A. CH_3CH_2^+
- B. $(\text{CH}_3)_2\text{CH}^+$
- C. $\text{CH}_3\text{CH}_2\text{CH}_2^+$
- D. CH_3C^+
- E. none of these

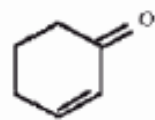
____ 4. Which compound would be expected to show intense IR absorption at 1680 cm^{-1} ?



A.



B.



C.

- A. A only
- B. B only
- C. C only
- D. A and C
- E. all of these

____ 5. What is the major organic product which results when cycloheptene is irradiated in the presence of *N*-bromosuccinimide?

- A. 1-bromocycloheptene
- B. 2-bromocycloheptene
- C. 1,2-dibromocycloheptane
- D. 3-bromocycloheptene
- E. 4-bromocycloheptene

6. When a high energy electron impacts molecule M in the ionization chamber, what type of species is initially produced?

- A. Cation B. anion C. radical D. radical cation E. radical anion

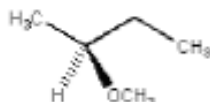
7. Which of the following most closely matches the C=C stretching frequency?

- A. 3300 B. 3000 C. 2200 D. 1700 E. 1200

8. Which of the following does not have a broad absorption with one or more spikes that is centered about 3300 cm^{-1} in the IR?

- A. $(\text{CH}_3\text{CH}_2\text{CH}_2)_3\text{N}$ B. $(\text{CH}_3\text{CH}_2\text{CH}_2)_2\text{NH}$ C. $\text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_2$
D. $(\text{CH}_3)_3\text{CNH}_2$ E. $(\text{CH}_2=\text{CHCH}_2)_2\text{NH}$

9. Predict the number of signals expected (disregarding splitting) in the ^1H NMR spectrum of the compound shown below.



- A. 6 B. 5 C. 4 D. 3 E. 2

10. What is the hybridization of the central carbon of allene (1,2-propadiene)?

- A. sp B. sp^2 C. sp^3 D. p E. none of the above

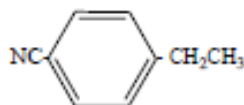
11. What descriptive term is applied to the type of diene represented by 2,4-hexadiene?

- A. conjugated diene B. cumulated diene C. isolated diene
D. alkynyl diene E. none of the above

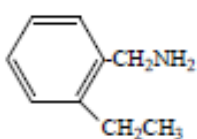
12. Provide a structure that is consistent with the data below. $\text{C}_9\text{H}_9\text{N}$

IR (cm^{-1}): 3050, 2950, 2240, 1630

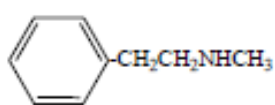
^1H NMR (d): 7.5 (2H, d), 7.1 (2H, d), 2.3 (2H, q), 0.9 (3H, t)



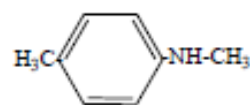
A.



B.

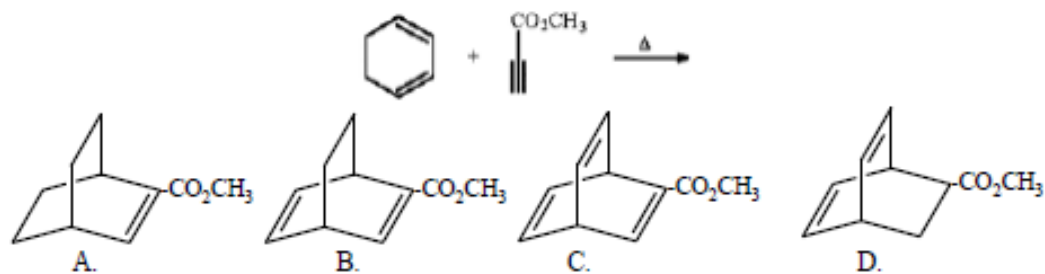


C.

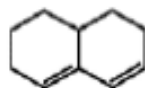


D.

13. Provide the structure of the major organic product in the following reaction.



14. Why does the diene shown below fail to undergo a Diels-Alder reaction with even the most reactive dienophiles?



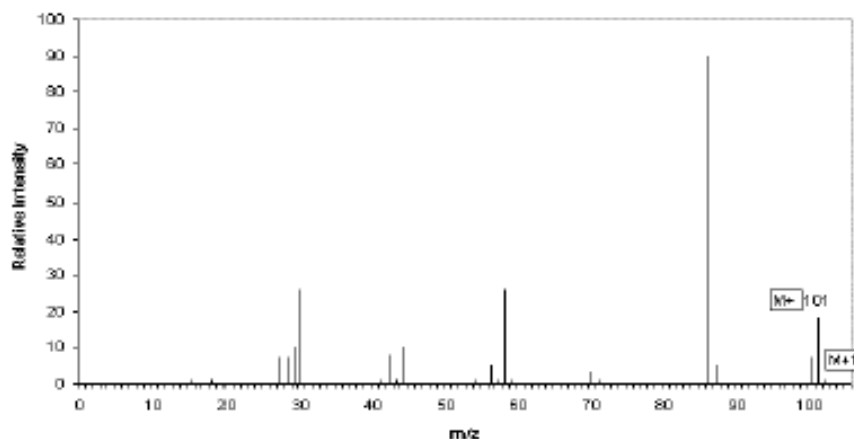
- A. The diene cannot achieve the necessary *s-cis* conformation. B. This is not a diene.
C. It is not monocyclic. D. It is a planar molecule. E. Do not obey Hückel's rule.

15. Which of the following is a correct description of benzene?

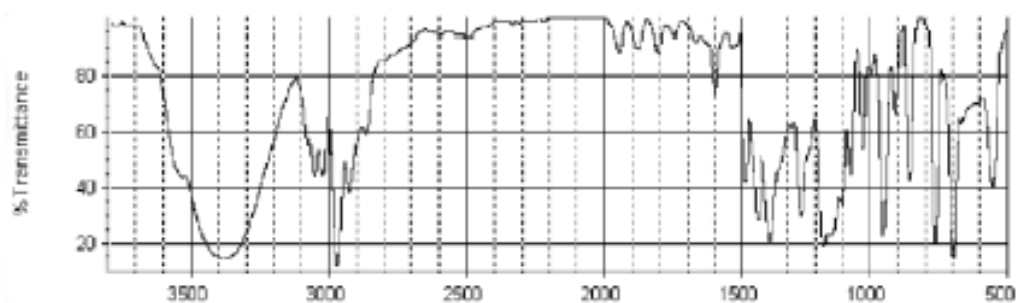
- A. The C-C-C bond angles are all equal to 120° .
B. Addition of Br_2 to benzene is highly unfavorable.
C. The molecule can be drawn as a resonance hybrid of two Kekulé structures.
D. All carbons are sp^2 hybridized.
E. All of these are correct.

Part II. Show all your work for complete credit.

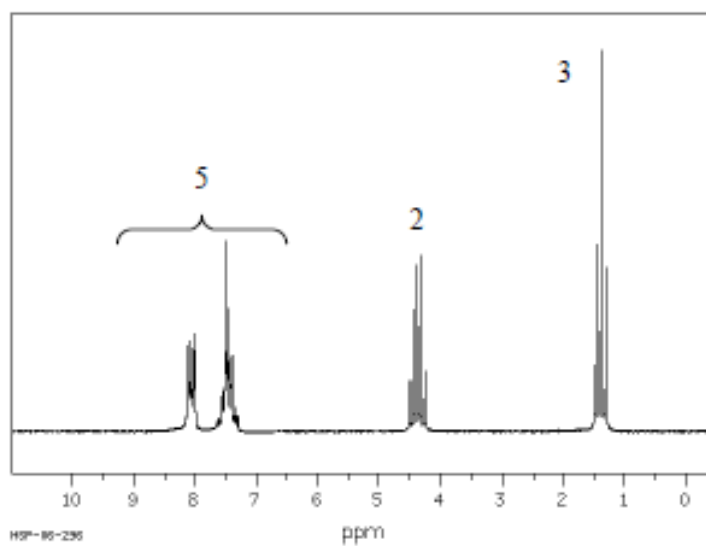
16. Suggest a structure, which is consistent with all the spectral data given below.
($\text{C}_6\text{H}_{15}\text{N}$)



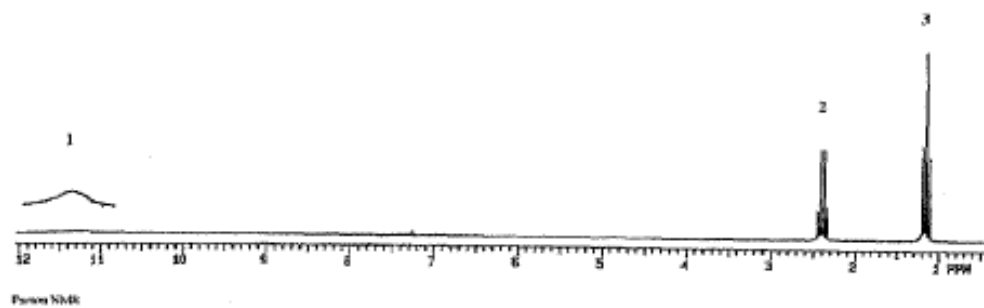
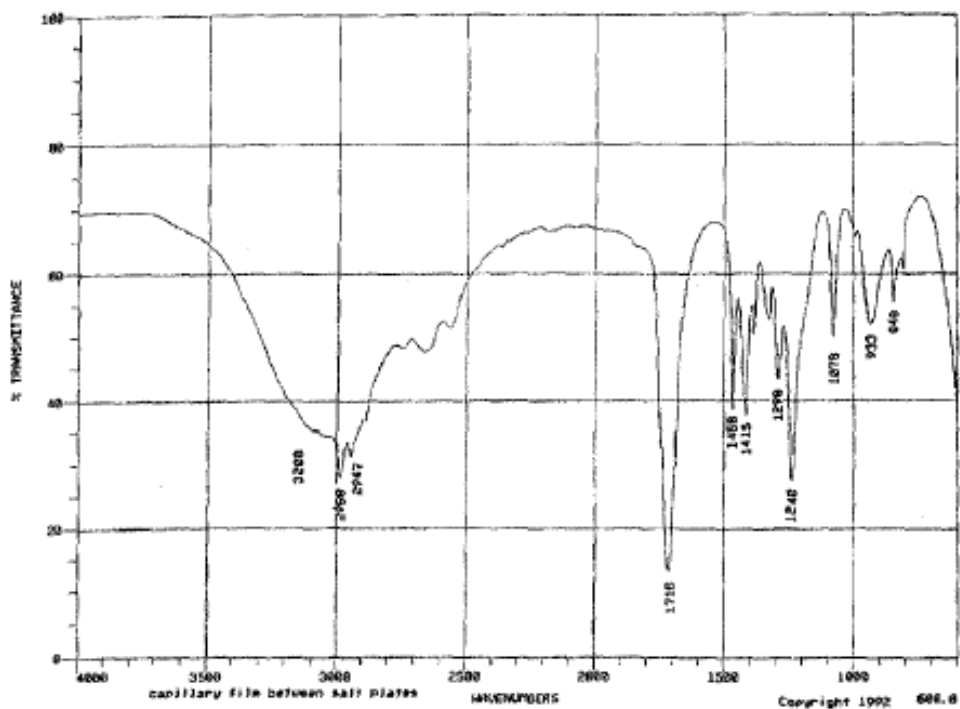
17. Suggest a structure, which is consistent with all the spectral data given below.
($C_9H_{12}O$)



18. Suggest a structure, which is consistent with the NMR spectra shown below. ($C_9H_{10}O_2$)



19. Suggest a structure, which is consistent with the IR and ^1H NMR spectra shown below. ($\text{C}_3\text{H}_6\text{O}_2$)



20. How many different type of hydrogens are there in the following compounds?

a) *m*-Chlorotoluene

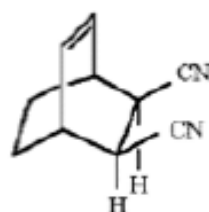
b) 1-Methyl-1,3-cyclohexadiene

21. Propose structure that fits the following ^1H NMR data:

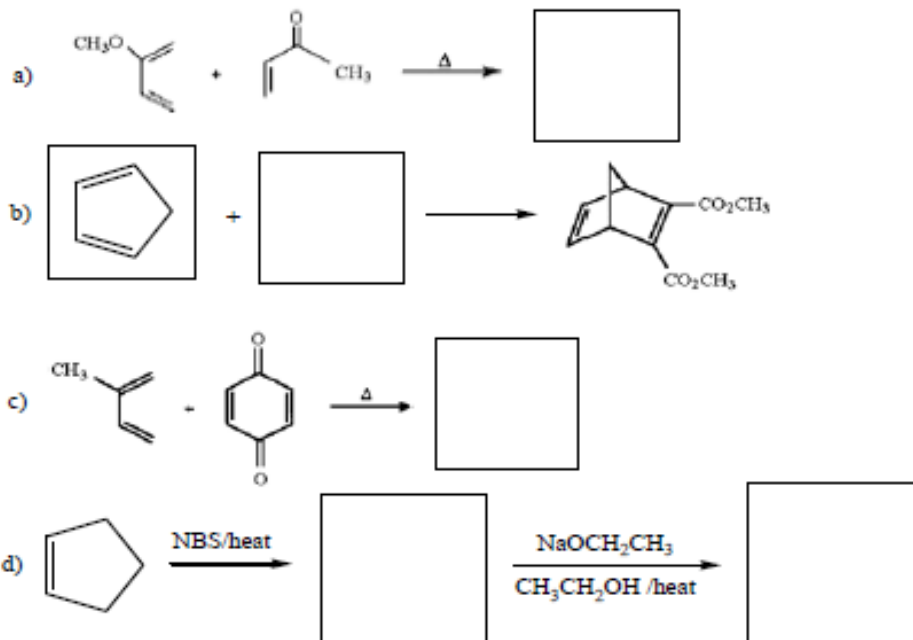
a) $\text{C}_3\text{H}_6\text{O}_2$ 3 signals: (3H, triplet); (2H, quartet); (1H, singlet)

b) $\text{C}_3\text{H}_8\text{O}$ 3 signals: (3H, triplet); (2H, quartet); (3H, singlet)

22. What diene and dienophile would you react to give the product below?

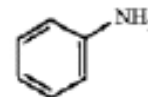
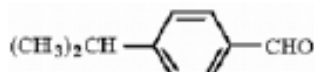
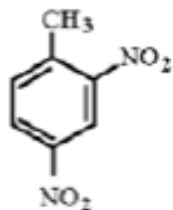


23. Provide the structures of missing compounds in the following reactions.



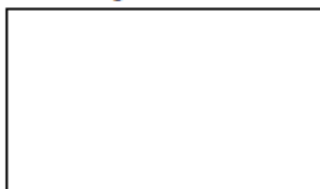
24. Nomenclature

I) Give the correct names (IUPAC)



II) Give the correct structures

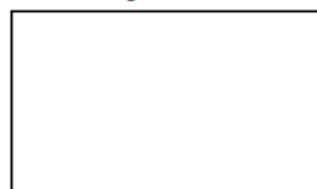
a) Acetophenone



b) 2-Bromo-4-chlorobenzoic acid

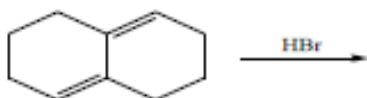


c) cis 1,3-pentadiene



Bonus question. Show all your work.

Provide a detailed stepwise mechanism for the reaction of 1,4 - addition of HBr to the diene shown below. Identify the major product and show all your work including electron flow.



Provide a detailed stepwise mechanism for the reaction of 1,2 - addition of Br₂ to the diene shown below. Identify the major product and show all your work including electron flow.

